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1924, Louis de Broglie used  
Einstein ' s equations to show that

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electrons can act like waves, just as photons can act like particles. 15.2.1  
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idea is known as de Broglie ' s hypothesis of matter waves. In 1926, De Broglie ' s hypothesis, together with Bohr ' s early quantum theory, led to the development of a new theory of wave quantum mechanics to describe the physics of atoms and subatomic particles.

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Louis de Broglie - Wikipedia  
Louis-Victor-Pierre-Raymond, 7e duc de Broglie was a French physicist who made groundbreaking contributions to quantum theory. In his 1924 PhD thesis he Solving the measurement problem: de Broglie-Bohm loses... 1 Feb 2008 A common attitude amongst

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De Broglie's phase wave and periodic phenomenon. De Broglie's thesis started from the hypothesis, " that to each portion of energy with a proper mass  $m_0$  one may associate a

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periodic phenomenon of the  
frequency  $\nu_0$ , such that one finds:  
 $h \nu_0 = m_0 c^2$ . The frequency  $\nu_0$  is  
to be measured, of course, in the rest  
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de Broglie equation states that a matter can act as waves much like light and radiation which also behave as waves and particles. The equation further explains that a beam of electrons can also be diffracted just like a beam of light. In essence, the de

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Broglie equation helps us understand the idea of matter having a wavelength.

de Broglie Equation - Derivation, de Broglie Wavelength ...

Abstract: Introductory course on quantum and atomic physics

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including optics and statistical physics. Objective: A basic introduction to quantum and atomic physics, including basics of optics and equilibrium statistical physics.

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When considering de Broglie wave

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phenomena of larger and more complex objects than atoms, fullerenes come to mind as suitable candidates. After their discovery<sup>9</sup> and the subsequent invention of efficient mass-production methods<sup>10</sup>, they became easily available. In our experiment (see Fig. 1) we use

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through walls. Thus a heavy atomic nucleus occasionally can eject a piece of itself in a process called alpha decay. The piece of nucleus (alpha particle) has insufficient energy as a particle to overcome the force barrier surrounding the nucleus; but as a wave it can leak through the

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barrier—that is, it has a finite probability ...

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