

## Introduction To Thermodynamics Prof Yu Qiaos Homepage

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ESS55 Prof. Jin-Yi Yu The First Law of Thermodynamics [This law states that (1) heat is a form of energy that (2) its conversion into other forms of energy is such that total energy is conserved. [The change in the internal energy of a system is equal to the heat added to the system minus the work down by the system:

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Ch 1 - Introduction: Basic Concepts of Thermodynamics. In this lesson, we will consider what thermodynamics is all about. We will also consider the concept of energy and discuss three types of energy: potential, kinetic and internal energies.

**Chapter 1 - Introduction: Basic Concepts of Thermodynamics**

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Thermodynamics is one of the most fascination branches of science. The laws of thermodynamics have wide applicability and are used in several branches of engineering and sciences. Being a quantitative subject, an exposure to a considerable number of problems will greatly enhance the reader's confidence and ability in applying the principles of thermodynamics for solving practical problems.

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Thermodynamics is a branch of physics that deals with heat, work, and temperature, and their relation to energy, radiation, and physical properties of matter. The behavior of these quantities is governed by the four laws of thermodynamics which convey a quantitative description using measurable macroscopic physical quantities , but may be explained in terms of microscopic constituents by statistical mechanics .

**Thermodynamics - Wikipedia**

Prof. Jin-Yi Yu The First Law of Thermodynamics • This law states that (1) heat is a form of energy that (2) its conversion into other forms of energy is such that total energy is conserved. • The change in the internal energy of a system is equal to the heat added to the system minus the work down by the system: U = Q -W

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To be able to use the First Law of Thermodynamics to estimate the potential for thermo- mechanical energy conversion in aerospace power and propulsion systems. Measurable outcomes (assessment method) : 1) To be able to state the First Law and to define heat, work, thermal efficiency and the difference between various forms of energy.

**THERMODYNAMICS: COURSE INTRODUCTION**

established principle of thermodynamics (which eventually became the Second Law) was formulated by Sadi Carnot in 1824. By 1860, as found in the works of those such as Rudolf Clausius and William Thomson, there were two established "principles" of thermodynamics, the first principle and the second principle.

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